Home Search Collections Journals About Contact us My IOPscience

On the blueshift in $Sn_{1-x}Co_xO_{2-\delta}$ transparent ferromagnetic semiconductor thin films

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 2007 J. Phys.: Condens. Matter 19 016216 (http://iopscience.iop.org/0953-8984/19/1/016216)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 129.252.86.83 The article was downloaded on 28/05/2010 at 15:03

Please note that terms and conditions apply.

J. Phys.: Condens. Matter 19 (2007) 016216 (7pp)

On the blueshift in $Sn_{1-x}Co_xO_{2-\delta}$ transparent ferromagnetic semiconductor thin films

K Gopinadhan, Dinesh K Pandya, Subhash C Kashyap and Sujeet Chaudhary 1

Thin Film Laboratory, Department of Physics, Indian Institute of Technology Delhi, New Delhi-110016, India

E-mail: sujeetc@physics.iitd.ac.in

Received 30 October 2006, in final form 16 November 2006 Published 7 December 2006 Online at stacks.iop.org/JPhysCM/19/016216

Abstract

Optical properties of spray pyrolysed $Sn_{1-x}Co_xO_{2-\delta}$ ($0 \le x \le 0.15$) thin films exhibiting room temperature ferromagnetism have been investigated in the energy range 1–6 eV. The onset of optical absorption is blueshifted by as much as 220 meV for a cobalt concentration of 10a/o (x = 0.10). Our detailed analysis indicates that the most probable cause of the blueshift is the Burstein– Moss effect. The optical conductivity calculations reveal that Co does not introduce any mid-gap states. Also the small change in the slope of α^2 versus $h\nu$ indicates that the O and Sn band edges are affected on Co addition. This bandedge shift also implies the presence of strong Coulomb interaction between Co, Sn and O sites, as a plausible cause of the observed ferromagnetism in these samples. The Burstein–Moss effect seems to be masking the effect of band structure modification introduced by Co addition. The high (~75%) optical transparency of these films in the visible region remains practically unaffected on Co addition.

1. Introduction

Tin oxide is one of the rare forms of semiconductors exhibiting high transparency and high electrical conductivity simultaneously. Due to this, tin oxide thin films have been widely used as transparent electrodes in LEDs, electrochromic displays, liquid crystal displays, solar cells, CCDs [1] etc. A potential application in the emerging field of spintronics [2] becomes possible when transition metal doping (Co, Mn, Fe etc) is done in SnO₂, and room temperature ferromagnetism (RTFM) is induced [3]. We have recently reported carrier mediated RTFM in Sn_{1-x}Co_xO_{2-\delta} thin films prepared by a spray pyrolysis technique [4]. In this paper, we report the optical properties of these well characterized ferromagnetic semiconducting thin

0953-8984/07/016216+07\$30.00 © 2007 IOP Publishing Ltd Printed in the UK

¹ Author to whom any correspondence should be addressed.

films. The study helps us to understand the origin of the observed blueshift in thin films of transparent ferromagnetic $Sn_{1-x}Co_xO_{2-\delta}$ semiconductor and its relation with the observed ferromagnetism. Generally, the blueshift in the optical absorption of semiconductors is attributed to many reasons, such as a decrease in lattice constant [5] (due to internal strain), the quantum size effect [6], electronic modification of the band edges [7] and the Burstein–Moss shift [8]. The present study of various physical parameters (aspects) of $Sn_{1-x}Co_xO_{2-\delta}$ thin films indicates that the observed blueshift may be attributed to the Burstein–Moss effect and electronic structural changes. The Burstein–Moss effect is due to the partial filling of the lowest states of the conduction band, which leads to a blocking of the lowest states, and hence an electronic transition from the valence band to the conduction band has to occur at higher energies, which result in the widening of the optically observed band gap.

2. Experimental details

Thin films of $Sn_{1-x}Co_xO_{2-\delta}$ with x = 0, 0.05, 0.10 and 0.15 were grown on quartz substrates at a temperature of 450 °C by spray pyrolysing a mixture of 1 M aqueous solutions of $SnCl_4.5H_2O$ and $(CH_3COO)_2Co.4H_2O$ in appropriate proportions. Spray depositions were carried out for 20–25 s in the ambient atmosphere using nitrogen as a carrier gas at a pressure of $\sim 4.0 \times 10^4$ N m⁻². The resultant films were annealed in air at 500 °C for 30 min to obtain a homogeneous and stable material. The homogeneity and stability of the films was confirmed by carrying out the deposition a number of times under identical conditions and observing the same physical parameters. The structural characterization of the films was made by using Philips X'Pert PRO (model PW 3040) diffractometer in the glancing angle (GA) mode with $GA = 2^{\circ}$. The goniometric resolution of the instrument is 0.001°. Hall measurements were carried out in the van der Pauw geometry at room temperature. Room temperature normal transmittance (T) and near-normal reflectance (R) measurements as a function of incident energy varying from 1 to 6 eV were carried out using a double beam spectrophotometer (Perkin Elmer, model Lamda 900) having a spectral resolution of 0.05 and 0.2 nm in the UV–VIS and NIR ranges, respectively. The complex refractive index $\tilde{n}(h\nu) = n + ik$ has been calculated numerically, by solving the Tomlin equations [9] for T and R. From the $\tilde{n}(h\nu)$ values thus obtained, it was possible to evaluate other optical constants, like the optical absorption coefficient $\alpha(h\nu)$ and the complex optical conductivity $\tilde{\sigma}(h\nu)$. The thickness of various films was measured to be (~350–460 nm) using a stylus profilometer (model XP-II, Ambios Technologies); this was then verified by calculating the thickness using the approximation of 'weakly absorbing films' [10].

3. Results and discussion

Figure 1 shows the transmittance curves of various $\text{Sn}_{1-x}\text{Co}_x\text{O}_{2-\delta}$ ($0 \le x \le 0.15$) thin films recorded under normal incidence at room temperature (RT). It may be noted that the transmittance in the visible region is not much affected on Co incorporation, which is technologically quite promising. The transmittance edge shows a blueshift with increase in the Co concentration up to x = 0.10 in the films. The blueshift can be quantified in many ways including extrapolation towards zero either in the plots of α versus $h\nu$ [11] or α^2 versus $h\nu$ [9], determination of the energy corresponding to 10% [12] or 50% [13, 14] transmittance through the samples or energy corresponding to a high value of α (10^5 cm⁻¹) [15]. In the present case, the observed blueshift has been quantified in terms of shift in the energy of incident radiation corresponding to 20% transmittance. This value has been chosen as it corresponds to the nearly linear rapid fall of the film transmittance and thus can be determined with fair precision. It may be noted from figure 1 that this energy value is maximum for x = 0.10, i.e. a shift of about



Figure 1. Normal transmittance T(%) versus incident photon energy $(h\nu)$ of $\operatorname{Sn}_{1-x}\operatorname{Co}_x\operatorname{O}_{2-\delta}$ films with $(0 \leq x \leq 0.15)$ deposited at 450 °C showing a blueshift with increase in Co concentration, *x*. Visible transmittance is not much affected on Co addition.



Figure 2. Normal transmittance T(%) as a function of incident photon energy $(h\nu)$ of the Sn_{0.95}Co_{0.05}O_{2- δ} films prepared at 450 and 350 °C, indicating that there is no blueshift on varying the crystallite size.

200 meV (which was later found to be a good estimation as shown in figure 3). At x = 0.15, the blueshift drops back to a lower value from its maximum value at x = 0.10. This can be understood as at x = 0.15 the film was found to contain not only SnO₂ but also the Co₃O₄ phase [4]. This indicates that in this film only a fraction of the Co ions is incorporated in the SnO₂ matrix and the rest are used in the formation of the segregated Co₃O₄ phase.

In order to look into the possible causes for the observed blueshift, we consider first the role of variation of the lattice parameters of SnO₂ with Co concentration. The x-ray diffraction (XRD) patterns of films with $0 \le x \le 0.10$ showed [4] peaks corresponding to the SnO₂ phase only. From the detailed XRD studies the lattice parameters *a* and *c* of the films having different concentration of Co have been calculated from the Bragg peaks after correcting for instrumental broadening (as shown in table 1). The data show that there is not much change in the lattice parameters for *x* varying up to 0.10. However, a tendency for a slight increase in *c* parameter is revealed. It suggests, if at all, a decrease in the optical band gap leading to a redshift. The small decrease of lattice parameter *a* in the case of films with x = 0.15 can be attributed to the presence of a Co₃O₄ phase in this film. The presence of Co₃O₄ is likely to introduce compressive strain in the SnO₂ crystallites thereby causing a reduction in lattice parameters. It can thus be concluded that the continuously increasing blueshift up to x = 0.10 is not due to the change in the size of the lattice.

Table 1. Variation of lattice parameters and crystallite size of $\text{Sn}_{1-x}\text{Co}_x\text{O}_{2-\delta}$ thin films with Co concentration for films deposited at 350 and 450 °C.

	Lattice constants (Å)		
Co concentration (x)	a	С	Crystallite size (nm)
0 (JCPDS data)	4.737	3.186	_
$0 (T_{\rm s} = 450^{\circ}{\rm C})$	4.737	3.184	26.7
$0.05 (T_{\rm s} = 450 ^{\circ}{\rm C})$	4.738	3.187	13.6
$0.10 (T_{\rm s} = 450 ^{\circ}{\rm C})$	4.737	3.186	12.3
$0.15 ({\rm SnO}_2) (T_{\rm s} = 450^{\circ}{\rm C})$	4.733	3.185	9.1
Pure Co ₃ O ₄ film	_	_	16.2
$0.05 (T_{\rm s} = 350 ^{\circ}{\rm C})$	4.711	3.168	10.1

To explore the possibility of a quantum size effect, a detailed analysis of the crystallite size from x-ray patterns has been carried out. As given in table 1, the crystallite size is seen to reduce continuously from 26.7 to 9.1 nm on increasing the Co concentration from x = 0 to 0.15. To see if the quantum size effect can account for the observed blueshift, we prepared samples of Sn_{0.95}Co_{0.05}O_{2- δ} at two different temperatures—450 and 350 °C—with a view to obtaining different crystallite sizes at fixed concentration. The crystallite size calculation now reveals the sizes to be 13.6 and 10.1 nm, respectively. The normal transmittance behaviour (figure 2) of these films grown at 450 and 350 °C shows that there is no appreciable change in the transmittance edge. This indicates that the origin of the blueshift due to cobalt incorporation is not associated with the quantum-size effect. Our data thus suggest that the Bohr excitonic radius for SnO₂ is smaller than 9.1 nm. Also, theoretically the expected Bohr exciton radius, given approximately by $a_{ex} = a_0 \varepsilon/m^*$, where a_0 is the Bohr radius, $\varepsilon \approx 13$ is the static dielectric constant and m^* is the exciton reduced mass that may be replaced by the electron effective mass (since the holes are very heavy $m_h \approx 3m_0$ in SnO₂ semiconductor and $m_e^* = 0.20m_0$ (present work)), is about 3.24 nm [16].

The optical band gap has been determined from the spectral dependence of the absorption coefficient $\alpha(h\nu)$, where $\alpha = \frac{4\pi\kappa}{\lambda}$, λ is the wavelength of the incident radiation and κ is the imaginary part of the complex refractive index (extinction coefficient). At photon energies exceeding the band gap (E_g) , the absorption coefficient α varies linearly with $(h\nu - E_g)^{1/2}$ for a direct gap semiconductor and with $(h\nu - E_g)^2$ for an indirect gap semiconductor. Band structure calculations [17] have revealed a direct band gap for SnO₂ just below the onset of indirect transitions. The plots of α^2 versus $h\nu$ for our films (shown in figure 3) reveal that all the films ($0 \le x \le 0.15$) have a direct band gap. The linear extrapolation to zero absorption gives a band gap value of 4.0 eV for undoped SnO₂ thin film. On increasing the Co concentration x, the band gap shifts to higher energies, showing a maximum blueshift of 220 meV for x = 0.10. In addition, there is no evidence for strong absorption at energies below the gap.

Though the absorption coefficient studies gives the onset of band transitions, study of the optical conductivity (in particular, the real part of the optical conductivity $\sigma_1 \propto 4\pi n\kappa/\lambda$, where n and κ are the real and imaginary parts of refractive index, respectively) is a better tool for comparing the predictions of band theory. Figure 4 shows the plots for the variation of the real part of the optical conductivity with incident photon energy. It may be noted that below the band edge the optical conductivity is almost invariant until the incident energy of radiation exceeds the optical band gap. The absence of in-gap conductivity indicates that the on-site Coulomb energy on adding another electron to the Co is very large [7].

To investigate further if any modifications of the electronic structure are introduced by the addition of Co, we determined the slope of the linear part of α^2 versus $h\nu$ (figure 4) and



Figure 3. α^2 versus $h\nu$ plots computed for Co concentrations x = 0, 0.05, 0.10 and 0.15. This gives a direct band gap value of 4.0 eV for a degenerate SnO₂ semiconductor.



Figure 4. The real part of the optical conductivity (σ_1) plotted as a function of photon energy $(h\nu)$ of $\text{Sn}_{1-x}\text{Co}_x\text{O}_{2-\delta}$ with $(0 \le x \le 0.15)$ showing that Co does not introduce any mid-gap states.

obtained its variation with Co concentration. The small change in the slope of α^2 versus $h\nu$ indicates that the conduction band edge (which is a result of 90% of Sn s content [18]) and the valence band edge (which is a result of O p content) are affected by addition of Co. The Coulomb interaction between Co, Sn and O sites provides a strong interaction, which could also partially account for the observed ferromagnetism as proposed in [7], in addition to the predominant contribution from the polarization of the free charge carriers. Also it seems that the large Burstein-Moss effect is just masking the effect of the small band structure modification introduced by addition of Co. If the O and Sn band edges are not affected by the substitution of Co, this blueshift should be understood in terms of the reduction of the interband oscillation strength upon dilution of Sn by Co. However, this scenario leads not to a shift in the band edges but a change in the slope of the α^2 versus hv plot, contrary to our observation. Further, as evidenced from the x-ray analysis, at the extreme solubility limits the phases are SnO2 and Co_3O_4 . If SnO_2 ($E_g = 3.92 \text{ eV}$) and Co_3O_4 ($E_g = 1.5 \text{ eV}$ [19]) form a solid solution, then, using Vegard's law, there should have been a redshift of ~200 meV instead of a blueshift. Hall measurements indicate that the majority carriers are electrons and their concentration increases monotonically with Co concentration up to x = 0.10, i.e. from 0.82×10^{19} cm⁻³ at x = 0.0to 6.09×10^{19} cm⁻³ at x = 0.10. The origin of conduction electrons upon Co doping is understood [20] on the basis of charge balance formulae for Co replacing Sn in the SnO_2 matrix, given as $(Sn^{4+})_{1-x}(Co^{2+})_xO^{2-}_{2-x}$ and/or $(Sn^{4+})_{1-x}(Co^{3+})_xO^{2-}_{2-x/2}$.



Figure 5. Straight line behaviour of the E_g versus $n_e^{2/3}$ plot for $\operatorname{Sn}_{1-x}\operatorname{Co}_x\operatorname{O}_{2-\delta}$ thin films indicating a Burstein–Moss optical blueshift.

Assuming that each Co substitution generates, on average, one electron, E_g should have been redshifted by an amount of approximately 242 meV from the value of SnO₂. Therefore, the rigid alloy (solid solution) concept does not explain the observed blueshift. Shinde *et al* also reported an increase in carrier concentration with Co doping in TiO₂ [21].

On the other hand, the effective energy gap for direct transitions in the doped material in the case of unperturbed conduction and valence band edges is given by

$$E_{\rm g}^0 = E_{\rm g0} + \Delta E_{\rm g}^{\rm BM},$$

where the Burstein-Moss shift is given by

$$\Delta E_{\rm g}^{\rm BM} = \frac{\hbar^2}{2m_{\rm vc}^*} (3\pi^2 n_{\rm e})^{2/3},\tag{1}$$

where E_{g0} is the direct band gap and n_e is the free electron density, with the effective reduced mass

$$\frac{1}{m_{\rm vc}^*} = \frac{1}{m_{\rm v}^*} + \frac{1}{m_{\rm c}^*},$$

where m_v^* and m_c^* represents the effective mass for the hole and electron, respectively.

Equation (1) predicts a shift in the energy gap proportional to $n_e^{2/3}$. A plot of $n_e^{2/3}$ versus E_g (figure 5) shows a linear relationship, indicating that the blueshift in our case is a result of the Burstein–Moss effect. The intercept of linear extrapolation of $n_e^{2/3}$ versus E_g plot on the energy axis yields a value of 3.92 eV for the band gap of a non-degenerate SnO₂ semiconductor, which is in agreement with the reported value [22]. Also, the simulated data using equation (1) show that the best-fit results yield a value of $0.2m_0$ for m_e^* , $3m_0$ for m_v^* , where m_0 is the rest mass of the electron. These values are in agreement with the earlier reported results [1, 18]. Here, it is worthwhile mentioning that we neglected the role of electron–electron and electron–impurity scattering, which may not be realistic. However, it does not affect the conclusions of the present study. It is worth mentioning that the Burstein–Moss shift has been reported in a SnO₂ degenerate semiconductor [23] whereas it was observed for the first time in Sn_{1-x}Co_xO_{2-\delta} thin films.

4. Conclusions

Thin films of $Sn_{1-x}Co_xO_{2-\delta}$ transparent ferromagnetic semiconductor prepared by spray pyrolysis show a blueshift in the optical gap with increase in Co concentration. A maximum

blueshift of 220 meV is seen for x = 0.10. The linear behaviour of α^2 versus $h\nu$ and the relationship of the direct band gap with carrier concentration indicate the Burstein–Moss shift to be the most probable cause of this blueshift. The best fit results of the (carrier concentration)^{2/3} versus optical energy gap relation yield a value of $0.2m_0$ for the electron effective mass and $3m_0$ for the hole effective mass. The optical conductivity calculations show that the Co ions do not introduce any mid-gap states. Also the small change in the slope of the α^2 versus $h\nu$ plot indicates that the band edges are significantly affected by Co substitution. This band-edge shift implies strong Coulomb interaction, which may be partially responsible for the observed ferromagnetism in these films. Also the Moss–Burstein effect seems to be just masking the effect of band structure modification introduced by the addition of Co. Significantly, the optical transparency in the visible region is not affected upon addition of Co.

Acknowledgments

We gratefully acknowledge Professor M Wuttig, RWTH, Aachen, Germany, for many fruitful discussions and the Ministry of Information Technology, Government of India, for financial support.

References

- [1] Chopra K L, Major S and Pandya D K 1983 Thin Solid Films 102 1
- [2] Wolf S A, Awschalom D D, Buhrman R A, Daughton J M, von Molnar S, Roukes M L, Chtchelkanova A Y and Treger D M 2001 Science 294 1488
- [3] Ogale S B, Choudhary R J, Buban J P, Lofland S E, Shinde S R, Kale S N, Kulkarni V N, Higgins J, Lanci C, Simpson J R, Browning N D, Sarma S D, Drew H D, Greene R L and Venkatesan T 2003 *Phys. Rev. Lett.* 91 077205
- [4] Gopinadhan K, Pandya D K, Kashyap S C and Chaudhary S 2006 J. Appl. Phys. 99 126106
- [5] Sekiya T, Ohta S and Kurita S 2001 Int. J. Mod. Phys. B 15 3952
- [6] Lippens P E and Lannoo M 1989 Phys. Rev. B 39 10935
- [7] Simpson J R, Drew H D, Shinde S R, Choudhary R J, Ogale S B and Venkatesan T 2004 Phys. Rev. B 69 193205
- [8] Hamberg I, Granqvist C G, Berggren K-F, Sernelius B E and Engström L 1984 Phys. Rev. B 30 3240
- [9] Tomlin S G 1968 J. Phys. D: Appl. Phys. 1 1667
- [10] Manifacier J C, Gasiot J and Fillard J P 1976 J. Phys. E: Sci. Instrum. 9 1002
- [11] Manifacier J C, Szepessy L, Bresse J F, Perotin M and Stuck R 1979 Mater. Res. Bull. 14 163
- [12] Köstlin I, Jost R and Lems W 1975 Phys. Status Solidi a 29 87
- [13] Haines W G and Bube R H 1978 J. Appl. Phys. 49 304
- [14] Steckl A J and Mohammed G 1980 J. Appl. Phys. 51 3890
- [15] Mizuhashi M 1980 Thin Solid Films 70 91
- [16] Munnix S and Schmeits M 1982 Solid State Commun. 43 867
- [17] Chiodini N, Paleari A, DiMartino D and Spinolo G 2002 Appl. Phys. Lett. 81 1702
- [18] Robertson J 1984 Phys. Rev. B 30 3520
- [19] Kim K J and Park Y R 2003 Solid State Commun. 127 25
- [20] Fayat J and Castro M S 2003 J. Eur. Ceram. Soc. 23 1585
- [21] Shinde S R, Ogale S B, Sarma S D, Simpson J R, Drew H D, Lofland S E, Lanci C, Buban J P, Browning N D, Kulkarni V N, Higgins J, Sharma R P, Greene R L and Venkatesan T 2003 Phys. Rev. B 67 115211
- [22] Shanthi E, Banerjee A and Chopra K L 1982 Thin Solid Films 88 93
- [23] Shanthi E, Dutta V, Banerjee A and Chopra K L 1980 J. Appl. Phys. 51 6243